

AD-A110 743 COLORADO STATE UNIV FORT COLLINS

COLORADO STATE UNIV FORT COLLINS

**CORRELATION BETWEEN
JAN 82 C W WILMSEN**

F/G 7/4

INSULATING PROP--ETC(U)
DAAG29-80-6-0143

UNCLASSIFIED

1

ARO-15511.6-EL

DAA629-80-C-0143

NL

END

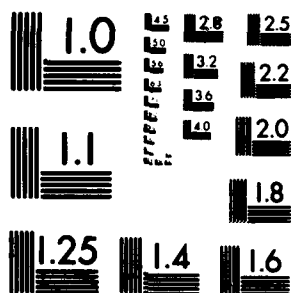
END
ONE

DATE
FILMED

3. 10.

DTIC

ONC



MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

AD A110743

DTIC FILE COPY

UNCLASSIFIED
SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

LEVEL II

ARO 75511.6-EL

12

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER 1	2. GOVT ACCESSION NO. AD-A110 743	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) Correlation Between the Composition/ Bonding and Insulating Properties of Grown Oxides on III-V Compound Semiconductors		5. TYPE OF REPORT & PERIOD COVERED Final Sept. 20, 1978-Sept. 30, 1981
7. AUTHOR(s) C. W. Wilmsen		6. PERFORMING ORG. REPORT NUMBER
9. PERFORMING ORGANIZATION NAME AND ADDRESS Colorado State University Fort Collins, CO 80523		8. CONTRACT OR GRANT NUMBER(s) DAAG29-78-G-0197 and DAAG29-80-G-0143
11. CONTROLLING OFFICE NAME AND ADDRESS U. S. Army Research Office Post Office Box 12211 Research Triangle Park, NC 27709		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		12. REPORT DATE January 6, 1982
		13. NUMBER OF PAGES 9
		15. SECURITY CLASS. (of this report) Unclassified
16. DISTRIBUTION STATEMENT (of this Report) Approved for public release; distribution unlimited.		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report) NA		
18. SUPPLEMENTARY NOTES The view, opinions, and/or findings contained in this report are those of the author(s) and should not be construed as an official Department of the Army position, policy, or decision, unless so designated by other documentation.		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) Oxides of III-V Compounds, InP, InAs, Insulators		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) This research investigated the composition, structure and electrical properties of thermal and anodic oxides of the III-V compounds. The composition of the oxides was determined by XPS profiling. Models for the structure of the oxides were obtained from current-voltage, X-ray diffraction and XPS data. The chemical composition and electrical properties of anodic oxides of InP were found to be highly dependent on the growth conditions but the oxides on GaAs and InAs were not. This appears to be related to the basic difference in the structure of the oxides. The anodic oxide could be formed in such a		

DTIC
ELECTE
FEB 10 1982
E

DD FORM 1473 EDITION OF 1 NOV 65 IS OBSOLETE
82 02 08 007

UNCLASSIFIED
SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE(When Data Entered)

20. ABSTRACT CONTINUED

way as to have good insulating properties. The thermal oxides were always poor insulators.

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A	



Unclassified

CORRELATION BETWEEN THE COMPOSITION/BONDING AND INSULATING
PROPERTIES OF GROWN OXIDES ON III-V COMPOUND SEMICONDUCTORS

Final Report

by

C. W. Wilmsen

January 1982

U. S. Army Research Office

DAAG29-78-G-0197
DAAG29-80-C-0143

Colorado State University
Fort Collins CO 80523

APPROVED FOR PUBLIC RELEASE;
DISTRIBUTION UNLIMITED

I. Problem Statement

This research investigated the composition, structure and electrical properties of thermal and anodic oxides of the III-V compounds. The composition of the oxides was determined by XPS profiling the grown oxides. Models for the structure of the oxides were obtained from current-voltage, X-ray diffraction and XPS data.

II. Summary of Results

A. Anodic Oxides. The anodic oxides of GaAs, InP and InAs were investigated. The compositional profiles of the InP anodic oxides were found to be highly dependent on growth conditions whereas the anodic oxides of InAs and GaAs were not. Similarly it was found that the insulating properties of the InP anodic oxides changed dramatically with growth condition while those of InAs and GaAs did not. For the InP oxides, Fig. 1 illustrates the large change in oxide resistivity with composition.

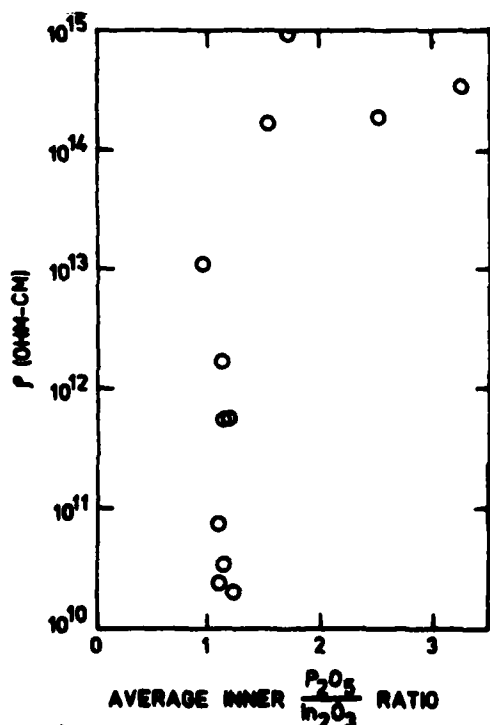


Fig. 1. InP anodic oxide resistivity at a field of 5×10^4 V/cm as a function of the average P_2O_5/In_2O_3 ratio of the inner oxide layer.

These observations led to the conclusion that the InP anodic oxide is phase separated into islands of In_2O_3 and P_2O_5 with the electrical conduction determined by electron tunneling or hopping between the islands. The anodic oxides of InAs and GaAs, on the other hand, do not appear to be phase separated and are more or less homogeneous. The island structure of InP anodic oxides seems to allow preferential dissolution of one or the other oxide species while the anodic oxides of InAs and GaAs tend not to dissolve selectively.

The insulating properties of the GaAs anodic oxides are approximately independent of growth conditions and the $\text{Ga}_2\text{O}_3/\text{As}_2\text{O}_3$ ratios are approximately equal to one for all cases. The InAs anodic oxides have similar characteristics but growth in basic solutions decreases the oxide resistivity. Table I summarizes the insulating properties of the anodic oxides.

Table I. Insulating Properties of the Anodic Oxides

III-V Compound	P, ohm-cm	Breakdown field, V/cm
InP	$10^{10} - 10^{15}$	$10^5 - 3 \times 10^6$
InAs	$10^{13} - 10^{15}$	$10^6 - 3.7 \times 10^6$
GaAs	$10^{15} - 10^{16}$?

The initial growth of the anodic oxides of GaAs and InP were also investigated. It was found that initial growth has a nucleation and island stage and that the oxide island of GaAs are different from those of InP. Photomicrograph of the GaAs anodic oxide islands are illustrated in Fig. 2.

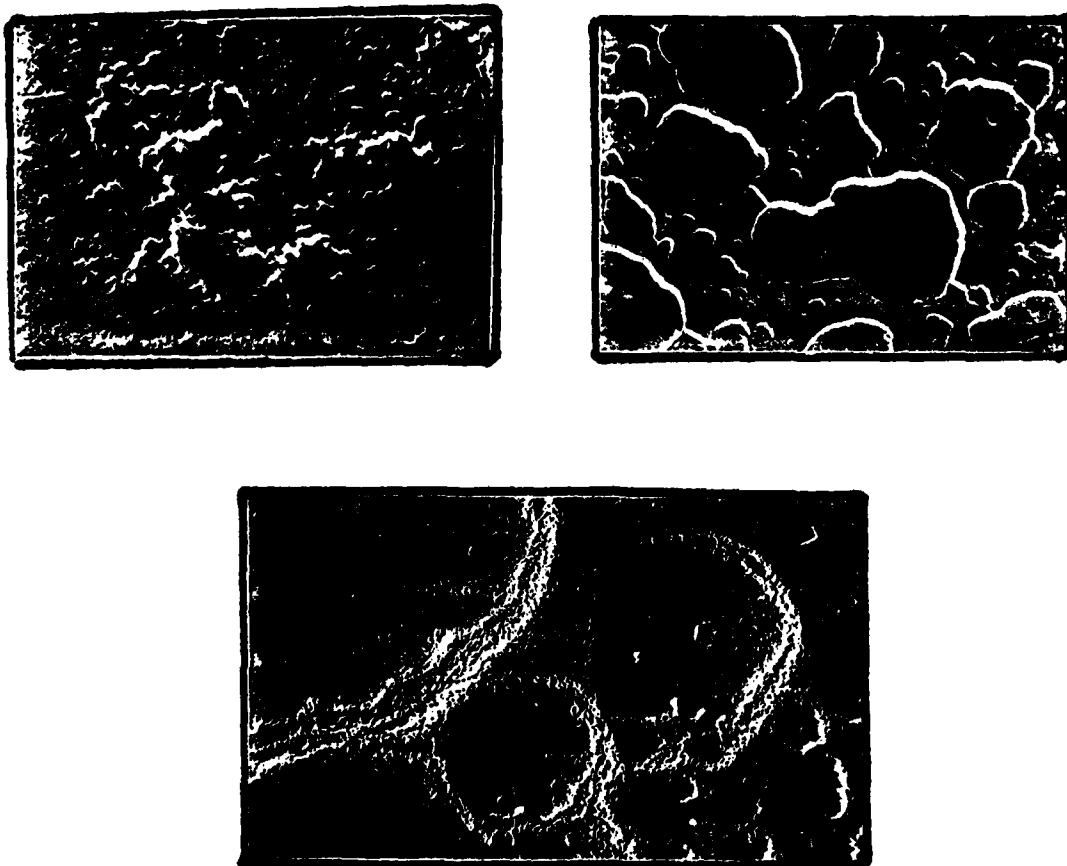


Fig. 2. Transmission electron micrographs of the nuclei and islands of anodic oxide on GaAs (Makky and Wilmsen).

From these and other photomicrographs it was determined that:

1. Nucleation is a continuous process
2. The island area and the island height are directly proportional up to an island area of $\approx 0.1 \mu^2$
3. Above $0.1 \mu^2$ the island height is independent of area
4. Coalescence occurs, probably by an increased growth rate in the neck regions and not by surface diffusion.

B. Thermal Oxides. The thermal oxides of InP, InAs and GaSb were investigated. Table II summarizes the results. The electrical Conduction of all these oxides was high.

III. List of Publications

1. "Thermal Oxidation of InAs," D. H. Laughlin and C. W. Wilmsen, Thin Solid Films, **70**, 325 (1980).
2. "Anodic Oxide Insulators on InP and InAs," D. A. Baglee, D. H. Laughlin, C. W. Wilmsen and D. K. Ferry in The Physics of MOS Insulators, edited by G. Lucovsky, S. T. Pantelides and F. L. Galeener, Pergamon Press, 1980.
3. "Inversion Layer Transport and Properties of Oxides on InAs," D. A. Baglee, D. K. Ferry, C. W. Wilmsen and H. H. Wieder, J. Vac. Sci. Tech., **17**, 1032 (1980).
4. "Inversion Layer Transport and Insulator Properties of Indium Based III-V's," D. A. Baglee, D. H. Laughlin, B. T. Moore, B. L. Eastep, D. K. Ferry and C. W. Wilmsen, Institute of Physics Conference Series Number 56, 1980.
5. "An Improved Anodic Oxide Insulator for InP Metal-Insulator-Semiconductor Field-Effect Transistors," D. H. Laughlin and C. W. Wilmsen, Appl. Phys. Lett., **37**, 915 (1980).
6. "Chemical Composition and Formation of Thermal and Anodic Oxide/III-V Compound Semiconductor Interfaces - Critical Review," C. W. Wilmsen, J. Vac. Sci. Technol., **19**, 279 (1981).
7. "Initial Anodic Oxide Growth on GaAs," W. Makky, F. Cabbara, K. Geib, and C. W. Wilmsen, submitted to J. Electrochem. Soc.

Table II. Composition and Crystallinity of the Thermal Oxides of InP, InAs and GaSb

III-V Compound	Growth Temp.	Composition	Crystallinity
InP	350-600°C	In_2O_3 and InPO_4 in a ratio of $\approx 4:1$ with elemental P at the interface	Non crystalline
	600-700°C	Primarily InPO_4 with no elemental P at the inter- face	Crystalline
InAs	350°C	In_2O_3 and As_2O_3 in \approx equal pro- portions	Non crystalline
	425-450°C	In_2O_3 and As_2O_3 in a ratio 3 to 1 with elemental As at the interface	Non crystalline
	475-500°C	$\text{In}_2\text{O}_3/\text{As}_2\text{O}_3 \approx 3$ with elemental In and As	Crystalline
GaSb	300-500°C	Ga_2O_3 and Sb_2O_3	Non crystalline

IV. Participating Scientific Personnel

C. W. Wilmsen - PI
Kent Geib - Research Associate
D. A. Baglee - Post Doc
Dennis Laughlin - M.S. August 1980
Art Nelson - M.S. Student (August 1982)
Wagih Makky - Ph.D. Student (December 1982).

END

DATE
FILMED

3-82

DTIC